

$$\partial(\sin^2 \theta) = -\mu/2 \cdot \sin^2 2\theta \cdot (1/\sin \theta + 1/\theta)$$

where μ is the systematic error constant for a particular film.

The program uses the relation

$$\sin^2 \theta \simeq \alpha A + \beta B + \gamma C + \partial(\sin^2 \theta);$$

hence the difference between the observed and theoretical values of $\sin^2 \theta$ is

$$\Delta = \sin^2 \theta - \alpha A - \beta B - \gamma C - \delta D.$$

By the method of least-squares the best values of A , B , C , D are those for which $\Sigma_i(w \cdot \Delta_i^2)$ is a minimum. The standard deviations of these quantities are also calculated.

For accurate work provision is made for inclusion of an eccentricity correction, which can be varied independently from the data.

Acta Cryst. (1967). **23**, 1114

Crystal data on some oxiodates. By MARINA BIAGINI CINGI, FRANCESCO EMILIANI and CARLO GUASTINI, *Istituti di Chimica Generale e di Mineralogia, Università degli Studi, Parma, Italy*

(Received 19 July 1967)

Crystal data and crystallographic descriptions are given of $\text{Sr}(\text{IO}_3)_2 \cdot \text{H}_2\text{O}$, $\text{Ca}(\text{IO}_3)_2 \cdot 6\text{H}_2\text{O}$, $\text{Sr}(\text{IO}_3)_2 \cdot 6\text{H}_2\text{O}$, $\text{CdH}_3\text{IO}_6 \cdot 3\text{H}_2\text{O}$, $\text{CaH}_3\text{IO}_6 \cdot 3\text{H}_2\text{O}$, $\text{Sr}(\text{IO}_4)_2 \cdot 7\text{H}_2\text{O}$.

In connexion with an investigation of the structures of the oxoanions of iodine(V) and iodine(VII) (Ferrari, Braibanti & Tiripicchio, 1965), the crystal data of some oxiodates of bivalent cations have been determined by X-ray Weissenberg methods (Cu $K\alpha$ radiation, $\lambda = 1.5418 \text{ \AA}$). The IV compounds were prepared by concentrating solutions of the corresponding periodates in concentrated hydrochloric acid and the VII compounds by evaporating solutions obtained by dissolving the metal carbonates in aqueous periodic acid. All the compounds are colourless. Densities were measured by the pycnometer method.

(1) $\text{Sr}(\text{IO}_3)_2 \cdot \text{H}_2\text{O}$. Stable, shiny, monoclinic crystals. Optically biaxial. Observed forms: $\{001\}$, $\{100\}$, $\{110\}$, $\{101\}$, $\{111\}$, $\{112\}$. Rotation axis for the X-ray analysis: $[010]$. $a = 13.12 \pm 1$, $b = 7.74 \pm 1$, $c = 8.95 \pm 1 \text{ \AA}$, $\beta_x = 133^\circ 6' \pm 10'$, $\beta_{\text{gon}} = 132^\circ 35' \pm 8'$, $V = 668 \text{ \AA}^3$, $Z = 4$, $D_m = 4.522$, $D_x = 4.520 \text{ g.cm}^{-3}$, $(a : b : c)_x = 1.695 : 1 : 1.156$, $(a : b : c)_{\text{gon}} = 1.709 : 1 : 1.160$. No piezoelectric effect was observed. Space group: $C2/c$ or Cc (from systematic absences). From the crystal data it appears to be a member of the isostructural series $\text{M}^{\text{II}}(\text{XO}_3)_2 \cdot \text{H}_2\text{O}$ ($\text{M}^{\text{II}} = \text{Ca}$, Sr , Ba , Pb and $\text{X} = \text{Cl}$, Br , I) (Groth, 1908).

(2) $\text{Ca}(\text{IO}_3)_2 \cdot 6\text{H}_2\text{O}$. Shiny, short piezoelectric prisms. Orthorhombic. Rotation axis: $[001]$. $a = 23.12 \pm 1$, $b = 15.03 \pm 1$, $c = 6.43 \pm 1 \text{ \AA}$, $V = 2236 \text{ \AA}^3$, $Z = 8$, $D_m = 2.967$, $D_x = 2.957 \text{ g.cm}^{-3}$. Space group: $Fdd2$ (from systematic absences). The X-ray data correspond to the goniometric measurements given by Marignac (1857), Eakle (1896) and later by Mélon (1935): $(a : b : c)_x = 1.538 : 1 : 0.428$, $(a : b : c)_{\text{gon}} = 1.547 : 1 : 0.428$ (from Mélon's data).

(3) $\text{Sr}(\text{IO}_3)_2 \cdot 6\text{H}_2\text{O}$. Small, prismatic, orthorhombic crystals. Rotation axis: $[001]$. $a = 23.57 \pm 1$, $b = 15.28 \pm 1$, $c = 6.50 \pm 1 \text{ \AA}$, $V = 2341 \text{ \AA}^3$, $Z = 8$, $D_m = 3.205$, $D_x = 3.097 \text{ g.cm}^{-3}$. Isostructural with (2).

(4) $\text{CdH}_3\text{IO}_6 \cdot 3\text{H}_2\text{O}$. Prismatic, monoclinic crystals which lose $4\text{H}_2\text{O}$ at 120° and the last $\frac{1}{2}\text{H}_2\text{O}$ at 170°C . Observed forms: $\{010\}$, $\{011\}$, $\{021\}$, $\{102\}$, $\{121\}$. Rota-

Data derived from Debye-Scherrer photographs and single-crystal rotation photographs are equally acceptable.

The print-out includes θ , observed d_{hkl} , calculated d_{hkl} , the weight, the value of the Nelson-Riley function, and the differences between observed and calculated $\sin^2 \theta$. Some variations are possible.

The program is written in I.C.T. 1900 Fortran (Fortran IV) for use on the 1905 computer. The listing together with full instructions for use are available from the authors on request. A similar, separate program has been written for the monoclinic system, for which a listing is available.

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tion axis: $[100]$. $a = 5.96 \pm 1$, $b = 14.38 \pm 1$, $c = 9.78 \pm 1 \text{ \AA}$, $\beta_x = 121^\circ 19' \pm 15'$, $\beta_{\text{gon}} = 120^\circ 25' \pm 11'$, $V = 723 \text{ \AA}^3$, $Z = 4$, $D_m = 3.605$, $D_x = 3.604 \text{ g.cm}^{-3}$, $(a : b : c)_x = 0.414 : 1 : 0.680$, $(a : b : c)_{\text{gon}} = 0.409 : 1 : 0.672$. No piezoelectric effect was observed in agreement with the space group ($P2_1/c$) deduced from the systematic absences. The crystals are probably the same as those studied by Rammelsberg (1869) and quoted in Groth (1908) as possibly orthorhombic.

(5) $\text{CaH}_3\text{IO}_6 \cdot 3\text{H}_2\text{O}$. Monoclinic prisms, which alter in air. Observed forms: $\{010\}$, $\{100\}$, $\{011\}$, $\{102\}$, $\{122\}$. Rotation axis: $[001]$. $a = 5.85 \pm 1$, $b = 14.56 \pm 1$, $c = 9.98 \pm 1 \text{ \AA}$, $\beta_x = 120^\circ 7' \pm 6'$, $\beta_{\text{gon}} = 120^\circ 5' \pm 12'$, $V = 736 \text{ \AA}^3$, $Z = 4$, $D_m = 2.920$, $D_x = 2.886 \text{ g.cm}^{-3}$, $(a : b : c)_x = 0.402 : 1 : 0.679$, $(a : b : c)_{\text{gon}} = 0.401 : 1 : 0.684$. Isostructural with (4).

(6) $\text{Sr}(\text{IO}_4)_2 \cdot 7\text{H}_2\text{O}$ or $\text{SrH}_2\text{I}_2\text{O}_9 \cdot 6\text{H}_2\text{O}$. Formula given on the basis of the chemical analysis. Thick, opaque, triclinic crystals. No piezoelectric effect was observed. Rotation axis: $[001]$. $a = 6.21 \pm 1$, $b = 11.15 \pm 1$, $c = 11.63 \pm 1 \text{ \AA}$, $\alpha = 55^\circ 25' \pm 8'$, $\beta = 73^\circ 35' \pm 1^\circ 23'$, $\gamma = 64^\circ 56' \pm 52'$, $V = 597 \text{ \AA}^3$, $Z = 2$, $D_m = 3.311$, $D_x = 3.309 \text{ g.cm}^{-3}$. The compound seems to correspond to that described by Rammelsberg (1868) as a hexahydrate.

Further structural work is in progress at this Institute for (1), (2) and (4) only.

Financial support for this research by the Consiglio Nazionale delle Ricerche (Roma) is gratefully acknowledged.

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